有機化學試題

本試題第1頁;共7頁

(如有缺頁或毀損 /應立即請監試人員補發)

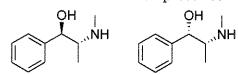
意一、本試題共二大題,共計100分。

思一二、請將最適當的答案依題號作答於考試答案卷上。 事一二、請將最適當的答案依題號作答於考試答案卷上。

事 三、試題答錯者不倒扣;題次號碼錯誤或不按順序或鉛筆作答,不予計分

I. Multiple choice. (毎題2%,共90%)

1. What is the relationship between the following two structures?



Enantiomers

Diastereomers

Constitutional Isomers

Identical

2. Which of the following structures is the following cation after rearrangement?



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→

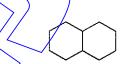
 $\bigoplus \oplus$

(H)

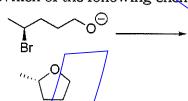
3. Which of the following chemicals is the product of the following Sn2 reaction?

NC NC

NC.



4. Which of the following chemicals is the product of the following Sn2 reaction?



 $\binom{\circ}{}$



5. Which of the following is the rate law for the following reaction?

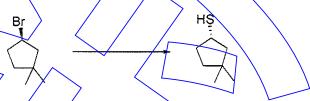
Rate = $k[H_2O]$

Rate = $k[H_2O][1$ -chloro-1-methylcyclohexane]

Rate = k[1-chloro-1-methylcyclohexane]

Rate = k[chloride ion]

6. Which of the following reagents could make the following reaction happened?



H2S in water

H₂S in DMSO

NaSH in water

NaSH in DMSO

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7. Which of the following reagents would best accomplish the following transformation?

HBr

NaBr

pyridine

pyridine then Nalkr

8. Which of the following is most likely to act as a base rather than a nucleophile?

_ Θ



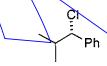
∑⊖ (

9. Which of the following alkyl halides would afford the indicated product upon reaction with sodium ethoxide?

Ph



Ph



10. The expected major product of the following reaction is:

Br



Br

Br

11. Which of the molecules below are enantiomers, formed as products upon reaction of HBr with 4-methylpent-1-ene?



II

II and III

III H Br

III and IV

H Br

II and IV

IV

12. The regioselectivity and stereospecificity in the oxymercuration-demercuration of an alkene is best described as:

Markovnikov orientation with syn-addition

Markovnikov orientation with anti-addition

anti-Markovnikov orientation with syn-addition

anti-Markovnikov orientation with anti-addition

13. What is the expected major product for the following reaction?

2. NaBH₄, NaOH

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14. What is the expected major product for the following reaction?

15. What major product is expected from the following reaction?

16. For the reaction sequence below, identify the expected major products.

17. What is the expected major product of the reaction shown below?

18. What is the expected major product for the following reaction?

19. For the reaction shown, select the expected major organic product.

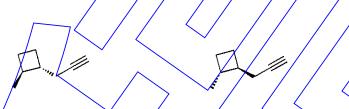


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本試題第4頁;共7頁

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20. For the reaction below, which of the alkynes listed would be expected to produce the product under the conditions shown?



21. Which sequence of reagents would be expected to accomplish the transformation of 1-butyne shown below?

- 1) H₂O/H₂SO₄, HgSO₄ cat.
- 2) H₂, Pt

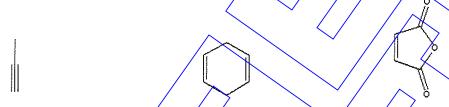
1) HBr

- 2) H₂, Pt
- 3) NaOH

- 1) H₂, Lindlar catalyst
- 2) BH3 THF/
- 3) H₂O₂, NaOH

- 1) HBr, peroxides
- 2) NaOH
- 3) H2, Pt

22. Which one of the following dienophiles is most reactive in the Diels-Alder reaction?



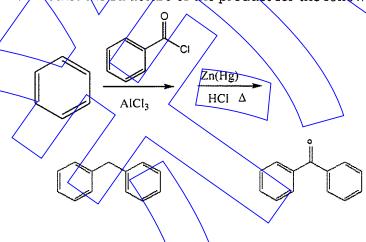
23. Which one of the following compounds is nonaromatic?

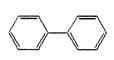


24. Predict the product for the following reaction.



25. Predict the structure of the product for the following reaction.





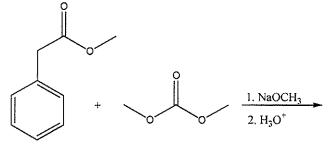


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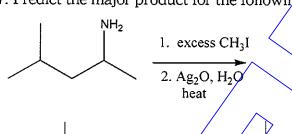
本試題第5頁;共7頁

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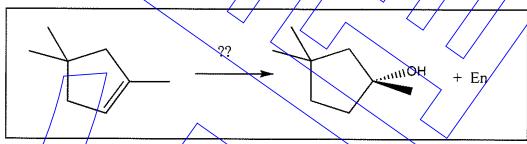
26. Predict the product for the following reaction.



27. Predict the major product for the following reaction.



28. Predict the reagent(s) required to complete the following transformation:



- 1) ØsO₄; 2) NaHSO₃, H₂Ø
- 1) Hg(OAc)2, H2O; 2) NaBH4
- 1) RCO3H; 2) H3O
- 1) BH3-THF; 2) H2O2, NaOH

29. Given that glucose has a specific rotation of +52.8°. Predict the concentration of a glucose aqueous solution contained in a 10 cm long polarimetry tube if a rotation of +15.8° was observed.

0.334 g/mL

0.299 g/mL

3.34 g/mL

0.0299 g/mL

30. Which of the following chemicals is a geminal dihalide?

isobutyl chloride

trans-1,4-diiodocyglohexane

cis-1,2-dibromocyclopentane

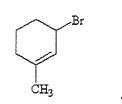
3,3-dichloropentane

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31. Rank the following molecules in order of increasing relative rate of SN1 solvolysis with methanol and heat (slowest to fastest reacting).



2 CH₃

3 CH₃

4 H₃ 5 H₃ B₁

2 < 3 < 4 < 5 < 1

3 < 2 < 4 < 5 < 1

2 < 3 < 4 < 1 < 5

5 < 4 < 3 < 2 < 1

32. Which of the following cycloalkenes would be expected to be stable?

2

3 🔾

2

1&2

1&3

2&3

33. Which of the following reagents should be used to convert hex-3-yne to (Z)-hex-3-ene?

H₂, Lindlar's catalyst

Na, NH₃

H₂, Pt

H₂SO₄, H₂O

34. Which of the following reagents or sequences do not produce an alcohol or diol from an alkene starting material?

HCO₃H

 OsO_4 , H_2O_2

Hg(OAc)₂, H₂O followed by NaBH₄

 H^{\dagger}, H_2Q

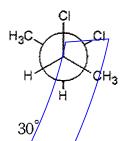
35. Which of the following reagents is the best choice for oxidizing a primary alcohol to an aldehyde? pyridinium chlorochromate (PCC)

LiAlH₄ (LAH)

Na₂Cr₂O₇, H₂SO₄

KMhO

36. For the compound represented by the Newman projection below, what is the dihedral angle between the methyl groups in the conformer that has the smallest molecular dipole moment?



<mark>/</mark>120°

120

60°

37. Treatment of tetrahydrofuran with excess HBr results in the formation of what major organic product?

1,2-dibromobutane

1-bromobutane

1,4-dibromobutane

1,5-dibromopentane

38. In biological systems, sulfonium salts such as SAM serve what function?

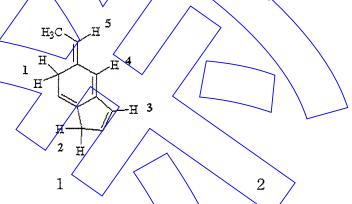
epoxidizing agents

esterifying agents

alkylating agents

crown ether agents

39. Which of the labeled H atoms (1-5) in the following molecule would be predicted to be the most acidic?



3

4

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40. The following compound has been found effective in treating pain and inflammation (J. Med. Chem. 2007, 4222).

Which sequence correctly ranks each carbonyl group in order of increasing reactivity toward nucleophilic addition?

1 < 3 < 2

1 < 2 < 3

2 < 3 < 1

3×1×2

41. The reagent which converts a carbonyl group of a ketone into a methylene group is

LiAlH₄

NaBH₄, CH₃CH₂OH

Zn(Hg), conc. HCl

Na, NH₃, CH₃CH₂OH

42. Physostigmine is used in the treatment of glaucoma. Within this structure, the atom indicated by _____ is most basic, while atom _____ is least basic.

1 (most basic), 4 (least basic)

1 (most basic), 3 (least basic)

2 (most basic), 3 (least basic)

2 (most basic), 4 (least basic)

43. Which of the following amines could be formed by reduction of an amide?

1. benzylamine

2. isopropylamine

3. aniline

4. triethylamine

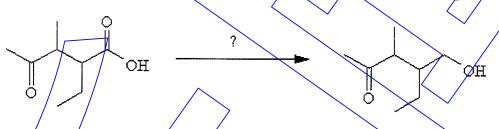
2 & 3

1, 3, & 4

3 & 4

1 & 4

44. Which of the reagents listed below would work best in the following reaction?



NaBH₄

LiAlH₄

LiAl[(OC(CH₃)₃]₃H

BH₃-THF

45. What compound is produced when N,N-dimethylpropanamide is treated with LiAlH.?

CH₂CH₂CH₂N(CH₃)₂

CH3CH2CONH2

CH₃CH₂N(CH₃)₂

CH₃CH₂CH₂NH₂

II. Filling the black, (10%)

1. Complete the following reaction by filling in the necessary reagents. (5%)

2)

HO₃S

2. Provide the structure of the major organic products which result from the reaction below. (5%)

CH₃ CH₂OH Δ