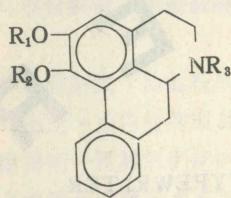


蓮之化學成分研究

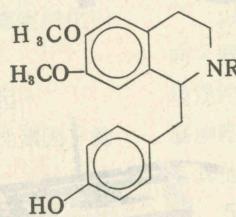
楊藏雄 陳繼明

蓮 (*Nelumbo nucifera* Gaertn., Nymphaeaceae) 原產於熱帶亞細亞為一多年生草本植物，台灣各地普遍栽植於水田邊或公園池塘，供觀賞或採食之用，其根莖（蓮藕）及種子（蓮子）有清涼，止渴，滋養，強壯之效；蓮房可治肺及腎臟之疾患；至於蓮葉及蓮柄則有抗菌，止血及解毒之作用兼治產後口渴，心煩諸症；蓮子蕊為蓮子中味苦之胚芽，有清心，去熱之效，可治勞心，吐血之疾。

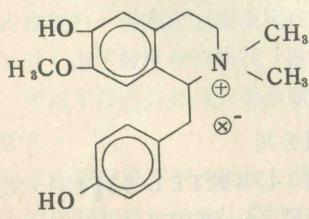
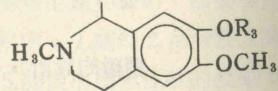
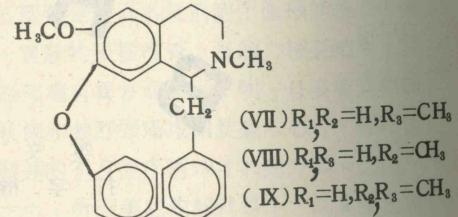
至於蓮之化學成分除蓮子含維生素丙，raffinose，胚芽含 asparagine，蓮藕含維生素丙，asparagine 之外，主要有生物鹼存在，關於 *Nelumbo* 屬植物生物鹼之化學研究，1899 年 Greshoff, Boorsma 等從蓮之子葉初次分離得一種物質曰 Nelumbine，但其性狀未詳，1959 年 Arthur 等從 Asiatic Lotus 之葉分離得 nuciferine (I)，其後日本富田等從日本產蓮單離出 nuciferine (I) 之外，還證明有 roemerine (II) 與 nornuciferine (III) 以及 dl-armepavine (VI)，1962 年，美 Kupchan 等由美國產同屬植物 *N. lutea*，除確認有 armepavine (VI)，nuciferine (I) 之存在，尙新單離出 L-(+)-N-norarmepavine (V) 及 (-)-N-norarmepavine (IV) 等生物鹼，同年高怡生等自中國大陸的蓮子蕊分離出 epavine (V) 及 (-)-N-norarmepavine (IV) 等生物鹼，同年高怡生等自中國大陸的蓮子蕊分離出 liesinine (VII)；而後我們從本省產的蓮子蕊單離兩種新成分，Isoliensinine (VIII) 與 lotusine (IX) 並證明蓮葉含 Nuciferine (I)，roemerine (II) 及 nornuciferine (III)，葉柄有 nuciferine (I) 的存在。1965 年日本古川由香港產的蓮子蕊分離另一種新物質 neferine (IV) 並證明有 nuciferine (I) 以及 pronuciferine (XI) 之存在；最近著者等更由本省的蓮子蕊再分離得 Neferine (IX) 與新的微量塩基性成分，命名為 methylcorypalline (XII)，後者經化學合成證明其構造為 6,7-dimethoxy-2-methyl-tetrahydroisoquinoline，這次主要關於新成分之單離方法，構造決定及合成概略介紹。



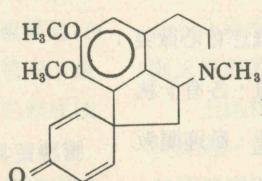
- (I) R₁, R₂, R₃=CH₃
 (II) R₁, R₂=CH₂, R₃=CH₃
 (III) R₁=H, R₂, R₃=CH₃
 (IV) R₁, R₂=CH₃, R₃=H



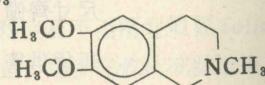
- (V) R=H
 (VI) R=CH₃



(X)

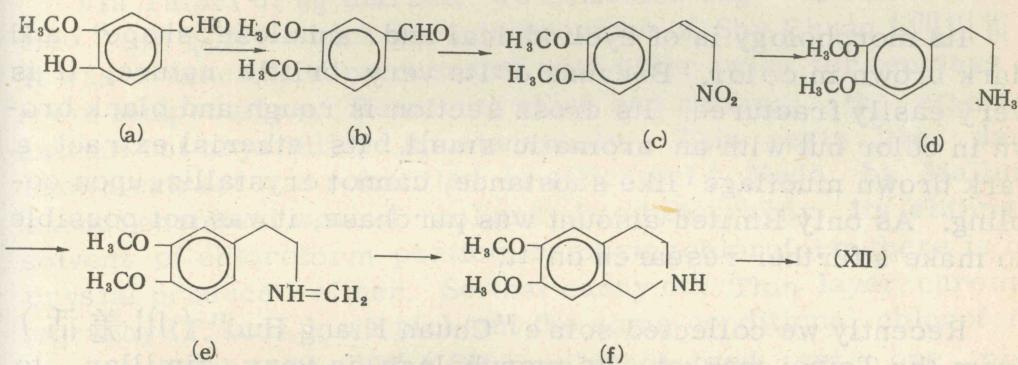


(XI)



(XII)

市售蓮子蕊以酒精抽取數次，其抽出液經減壓濃縮得黑色油狀物，以 3% 的醋酸溶液溶解，過濾之濾液，用乙醚振盪除去非塩基性物，酸性層加氨水鹼化後，析出的生物鹼以氯仿抽取，合併氯仿抽取層，加 NaOH 溶液分離酚性及非酚性塩基，氯仿層用酸鹼處理，水洗，脫水乾燥得精製非酚性的塩基，經 Al_2O_3 Column Chromatography 從 Benzene elution 得粗結晶，用 N-hexane 再結晶得白色羽狀結晶，量微 m.p. $58\sim60^\circ$, T.L.C. 1 Spot, $[\alpha]_{D}^{20}, \pm 0$, $\text{FeCl}_3(-)$, Labat (-), 塩酸塩 mp. $216\sim217^\circ$, 苦味酸塩 mp $152\sim153^\circ$, Methiodide, mp. 242° 。由 IR Spectrum 顯示有 $-\text{OCH}_3$ 基及 $1,2,4,5$ -tetrasubstituted benzenoid, UV 之極大吸收 (λ_{max}) 在 $281\text{m}\mu$, $285\text{m}\mu$, $290\text{m}\mu$ (Sh.), Mass Spectrum $M^+=207$, NMR Signals 在 6.18τ (6H, $2\times\text{OCH}_3$), 7.582 (3H, $-\text{NCH}_3$), 3.40 及 3.48τ (2H, Aromatic H) 綜合以上的性質，推定為 dimethoxy, N-methyltetrahydro-isoquinoline derivatives，本結晶性物質直接與我們合成如下之 6,7-dimethoxy-2-methyltetrahydroisoquinoline 以 IR, T.L.C. 比較結果完全一致，且 m.p. 混融不下降，此生物鹼之 $-\text{HCl}$, picrate, methiodide 等衍生物與合成品之諸衍生物 mp 混融亦一致，故證明由蓮子蕊抽取的塩基之構造式為 (XII)，因為構造上是 Manske (1937) 所分離的 Corypalline 之 7-O-Methyl 衍生物，故命名為 Methylcorypalline (XII)。



Vanillin (a) 以 dimethylsulfate methylation 得淡黃色針狀結晶，veratraldehyde (b) m.p. $40\sim40.5^\circ$ 。 (b) 於甲醇中與 nitromethane 及微量之 methylamine 室溫反應得黃色片狀結晶之 3,4-dimethoxy- ω -nitrostyrene (c), mp $139\sim140^\circ$ 。 (c) 之溶液中，以 lithium Aluminum hydride 還原得白色結晶之 3,4-dimethoxy phenylethylamine Oxalate (d) mp $178\sim179^\circ$ 。由 (d) oxalate 製成游離的塩基，加 formalin 溶液 (37%) formylation 反應得油狀物 (e)，再於酸中 Cyclization 得 6,7-dimethoxytetrahydroisoquinoline - HCl (e) 結晶，mp $254\sim255^\circ$ 。 (e) 化成游離塩基，以甲酸 (85%) 及 formalin (37%) 反應，經 hydroxymethylation 得白色羽狀結晶之 6,7-dimethoxy-2-methyltetrahydroisoquinoline (XII), mp $58\sim59^\circ$, HCl，白色顆粒狀結晶 mp. $214\sim5^\circ$; picrate，黃色微針狀結晶，mp $152\sim3^\circ$; methiodide，白色片狀結晶，mp $243\sim4^\circ$ 。

Methylcorypalline 的合成雖 Buck 曾製備過，但由自然界單離發現，今次為首次，此物質與 bisbenzylisoquinoline, benzylisoquinoline, aporheine 等生物鹼共存於蓮子蕊，在生合成觀點上，methylcorypalline 很可能為合成這些更複雜生物鹼之前驅物質 (biogenetic precursor) 或者為它之代謝分解物，1953 年 C. Hanna 和 J.H. Shutt 報告 substitute isoquinoline 衍生物包括 6,7-dimethoxy-2-methyltetrahydroisoquinoline (methylcorypalline) (XII) 類似 Papaverine 之作用，具有擴張冠狀動脈的效能。

綜合以上所述，蓮之生物鹼分佈在化學上可分為 (1) aporheines: (I), (II), (III), (IV), (2) benzyltetraisoquinolines: (V), (VI), (X) (3) bisbenzyltetra hydroisoquinoline: (VII), (VIII), (IX) (4) apomorphine protobase: (X), (5) tetrahydroisoquinoline: (XI) 等五類；於蓮葉柄、胚芽各部所含之成分亦有所差異，葉含 (I)~(IV); 葉柄含 (I), (II), 蓮子蕊有 (VII)~(XII)，故以植物分類學及生理學上與化學構造互相關係的觀點而言，蓮之生物鹼成份的差異頗具興趣的。

A STUDY ON
THE CHEMICAL CONSTITUENT OF CHINESE HERB
" KIANG HUO"

KUN YING YEN AND LING LING YANG
DEPARTMENT OF PHARMACEUTICAL PHYTOCHEMISTRY
TAIPEI MEDICAL COLLEGE

"Kiang Huo" (羌活) was originally grown in China. In former times, Dr. Kimura and Nakao¹ ascertained that it belonged to the *Angelica* genus (*Umbelliferae*). Author was studying at Kyoto University (Japan), we purchased some amount of "Kiang Huo" from a Chinese herb drug market in Osaka for the study of its chemical constituents.

Its morphology is of cyclindrical and annulose shape and dark brown in color. Because of its very brittle nature, it is very easily fractured. Its cross section is rough and black brown in color but with an aromatic smell. Its etherial extract, a dark brown mucilage like substance, cannot crystallize upon cooling. As only limited amount was purchase, it was not possible to make a further research on it.

Recently we collected some "Chuan Kiang Huo" (川羌活) from the Taipei market. Its morphology is very simillar to that of "Kiang Huo". In order to give a systematic study of the umbelliferous drugs, a further research on it was carried out once more.

As the etherial extract did not crystallize upon cooling, it was submitted to silica gel column chromatography. Then by applying the Thin layer chromatography (T. L. C.), we collected the same portion of chloroform and removed the chloroform. But no crystal produced either. However, when treated by saponification as shown in chart I, a crude crystal was obtained. After recrystallization with ethyl alcohol, its melting point raised to 189-190°C. The composition of the white needle crystalline (I) is $C_{12}H_{18}O_4$. When the Infra-red spectra of (I) were tested at 1725 cm^{-1} , it showed lactone and carbonyl group.

Based on the foregoing-mentioned melting point, composition and IR spectra, it is presumed that (I) is Bergapten when it was mixed with authentic sample, its melting point did not depress. This fact further strengthened our conviction that (I) is

Bergapten.

Bergapten is isolated from this herb drug by saponification. Ordinarily, *Angelica* genus posses blue fluorescent spots. It shows that this plant contains Umbelliferone. However, merely based on the above-mentioned factors, it still can hardly be determined whether it belongs to the family of *Angelica* genus or not.

Apart from Bergapten, we still continue to study very deliberately other substances.

Experimental Section

Isolation of Bergapten:

Via Taipei drug market, we collected 5kg "Chuan Kiang Huo" (川羌活) which grow in province of Shy Chuan (四川省). It was crushed and mercerated with ether twice for ten days at room temperature. Removing ether and cooling, the etherial extract not crystallized spontaneously. This resin like dark brown extract was submitted to silica gel (Made by Mallinckrodt chemical works) column chromatography. Its eluting solvent of chloroform portion, removing chloroform, there is no crystal produced either. So that carry out Thin layer chromatography (T. L. C.), we pick out the same conditional chloroform portion which add 10% KOH in alcohol and set one night for saponification. Then added enough water and removed alcohol by vaccum. Shaking with ether to remove the unsaponification portion.

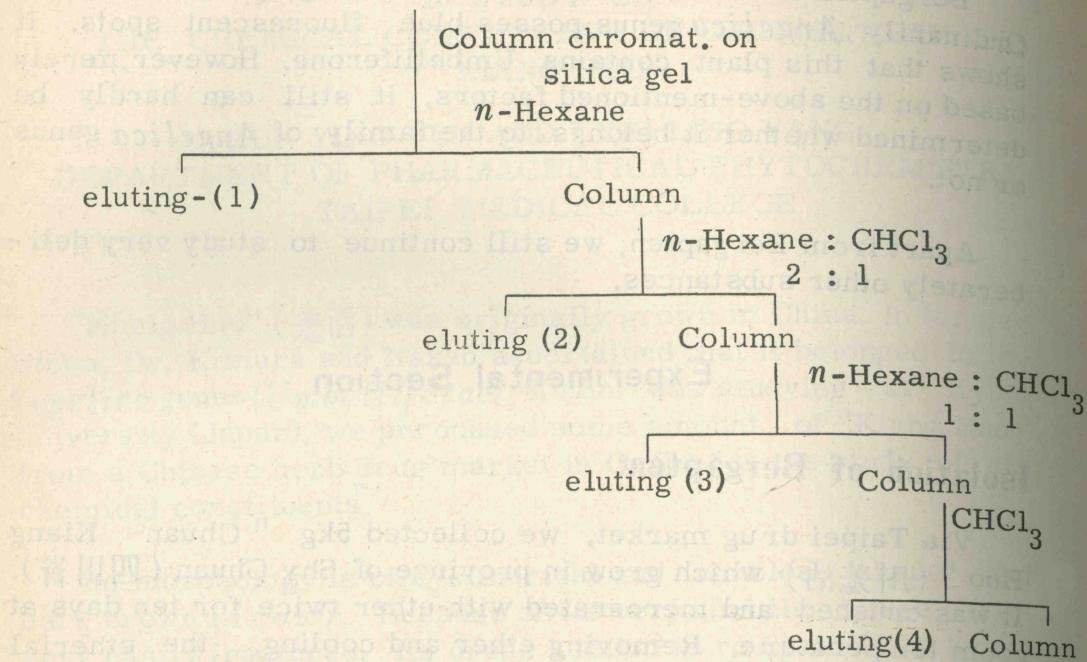
The water layer, acidified by dil. H_2SO_4 and shake out the coumarin fraction by ether. This ether solution is shaken by the following solution: saturated $NaHCO_3$, 5% $NaOH$. Removing the acidic portion and phenolic portion.

In the neutral portion, dehydration and removing ether, a crude crystall obtained. It treated recrystallization with ethanol, a white needle crystal, mp. $189-190^{\circ}C$ (I).

$C_{12}H_{18}O_4$ Anal. Calcd : C, 66.67% ; H, 3.70%
 Found : C, 66.80% ; H, 3.80%

The I R spectrum of (I) was identical and exhibited characteristic peak at 1725 cm^{-1} assignable to lactone, carbonyl group. Mixed with authentic sample of Bergapten no depressed melting point.

Chart I. Etherial extract of "Chuan Kian-Huo"



Saponification the same chloroform portion

evaped. + 10% EtOH-KOH (at room temp. for 1 day) + H₂O; EtOH evap in vacuum. Et₂O add.

aq. layer Et_2O layer
 acidify c 30% H_2SO_4 + 5% NaOH soln.
 + Et_2O add

acidic portion

Environ Biol Fish (2007) 80:31–39

layer

SO_4

| Et₂O add.

+Et₂O add

acidic portion

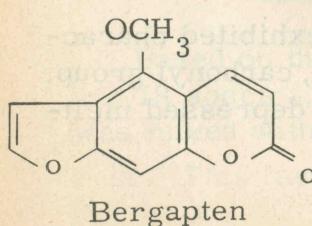
aq. layer

acidify c 30%

H₂SO₄

11250 4

add Et₂O



用藥之土壤學

References

- 1). M. Nakao, K. Kimura: The Annual Proceeding of Shanghai Natural Science Laboratory Vol. 1, No. 2, 88 (1929).
- 2). K. Kimura, K. Hata, K. Yen, S. Cheng: Yakugaku Zasshi (Tokyo) : 78, 442 (1958).

〔中文摘要〕

羌活之成分研究（I）

臺北醫學院 生藥化學科

顏焜熒 楊玲玲

臺北市生藥市場收集之大陸產“川羌活”5公斤，用乙醚(Ether)溫浸抽取，移去乙醚通 Silica gel Column Chromatography，展開劑順次用 *n*-Hexane, *n*-Hexane : CHCl₃ (2:1), *n*-Hexane : CHCl₃ (1:1), CHCl₃ oney。將 CHCl₃ 部分用 T.L.C 法將具有有相同 Spot 之部分合併，皂化後得 mp. 189 ~ 190°C，組成 C₁₂H₁₈O₄ 之白色針狀結晶，由紅外光譜 (IR) 及與標品混融，融點無下降，證實此結晶為 Bergapten。

(台灣藥學雜誌 Vol. 20 投稿中)